



The Effect of Mechanical Cycling on the Electronic Conductivity of Composite Electrodes

T. Brendel, M. Janzen, M. Müller, W. Bauer, D. Kramer, R. Mönig

Motivation

During battery operation, active materials change their volume. This can cause mechanical damage in the form of cracked active particles [1]. Even when the active material is not damaged, binder joints can move and carbon black (CB) can redistribute [2]. Here we explore how this affects the electrode performance. In order to isolate these mechanical effects, we exclude electrochemical changes by pure mechanical cycling. This serves as an experimental tool to determine changes of the conductive network and can be used as a test of the mechanical reliability. [1] Chen, D. et al. Electrochemical Acta 2018, 259, 939-948. DOI: 10.1016/j.electacta.2017.10.179



Variation of T

- Dry electrode sheet placed between two metal plates
- Force is chosen to cause similar stresses than during battery operation (as estimated by *in situ* substrate curvature experiments [2])
- · Simultaneous measurement of resistivity



[2] Janzen, M. et al. Energy Technology 2021, 9(6), 2000867. DOI: 10.1002/ente.202000867

Results 1. Comparison of Binder Materials

- Binder interconnects particles and binder with higher modulus stiffens the composite electrode
- Also increased amount of binder (from 5% to 10% PVDF) leads to a stiffer electrode
- Choice of binder material and amount allows tuning of electrode stiffness



Results 2. Electrode Rearrangement by Mechanical Cycling · Reduction of hysteresis of electronic ing strain /% resistance (R) and strain Reaches almost stress-independent R Different timescales for changes of R and thickness Particles move earlier than CB CB seems to escape from highly stressed regions Engineering strain (c) (d) 1.0 0%/ U 1.5 AR/Strain / 2 150 Applied 100 (b) AR/O Displacement /µm Cycle Numbe **Results 3. Thermal Activation** During cycling, R and thickness evolve asymptotically Increasing T initiates new rearrangement and leads to smaller values . Possibly caused by softening/plasticity of polymeric binder Athermal processes also cause changes in the conductive network

[3] but are not in agreement with our findings. 111.0 RT 95-L RT+38 ~vcle 201+ RT+18 Cycle 101+ 110.5 <u>a</u> 90 RT+18 E 110.0 RT+38 Mean 85 109.5 stance N 80 Distance 109.0 100 cycles 108.5 Resi 108.0 00 cvcle FP 5wt%PVDF 65 Cycle 99 107.5 0.5 1.0 1.5 3.5 4.0 4.57.0 7.5 8.0 Time [h] 1188011900 2534025360 1360 1380

[3] Becker, V. et al. Energy Technology 2021, 9(6), 2000886. DOI: 10.1002/ente.202000886

Conclusion

- Mechanical cycling leads to a rearrangement of the electrode. This was observed in the mechanical strain data, its hysteresis and in the electronic conductivity.
- During cycling, mechanical behavior and electronic resistivity depend on temperature. It can be expected that complex mechanical changes occur when a battery is operated at different temperatures.

Contact details

Thimo Brendel(thimo.brendel@kit.edu)Reiner Mönig(reiner.moenig@kit.edu)Dominik Kramer(dominik.kramer@kit.edu)



Next steps

- Influence of solvents on mechanical and electronic properties
 - Fatigue testing of electrodes by fast mechanical cycling



This work was supported by the Deutsche Forschungsgemeinschaft (DFG) through the Research Training Group GRK 2218 SiMET (Simulation of Mechanical, Electrical and Thermal processes in lithium-ion batteries)

www.iam.kit.edu